

This is the peer reviewed version of the following article:

Assessment of the basis set effect on the structural and electronic properties of organic-protected gold nanoclusters / Muniz Miranda, Francesco; Menziani, Maria Cristina; Pedone, Alfonso. - In: THEORETICAL CHEMISTRY ACCOUNTS. - ISSN 1432-881X. - ELETTRONICO. - 135:4(2016), pp. 1-9. [10.1007/s00214-016-1856-2]

*Terms of use:*

The terms and conditions for the reuse of this version of the manuscript are specified in the publishing policy. For all terms of use and more information see the publisher's website.

28/04/2026 15:39

(Article begins on next page)

[Click here to view linked References](#)

1  
2  
3  
4  
5  
6  
7  
8  
9 **Assessment of the Basis-Set Effect on the Structural**  
10  
11 **and Electronic Properties of Organic-Protected Gold**  
12  
13  
14  
15 **Nanoclusters**  
16  
17  
18  
19

20 Francesco Muniz-Miranda,\* Maria Cristina Menziani, and Alfonso Pedone  
21

22  
23 *University of Modena and Reggio Emilia (UniMoRE), Department of Chemical and Geological*  
24  
25 *Sciences (DSCG), Via G. Campi 103, I-41125, Modena, Italy*  
26  
27

28 E-mail: francesco.munizmiranda@unimore.it  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59

---

60 \*To whom correspondence should be addressed  
61  
62  
63  
64  
65

## Abstract

Here we have investigated the structural and optical properties of five monolayer-protected gold nanoclusters with a combination of exchange-correlation functionals, namely B-PBE for the geometry relaxation and CAM-B3LYP for the time-dependent calculations.

We have tested the accuracy of five different basis sets in reproducing the experimental structures of these nanoclusters, and we have found that even a rather small basis set (single zeta) can outperform a significantly larger one (double zeta) if some selected atoms are treated with polarization functions. Namely, the sulfur and phosphorous atoms of the capping thiols and phosphines usually are hypervalent when bonded to the gold inner core, therefore polarization functions allow them significantly more structural flexibility. With the two best performing basis sets we carried out optical calculations and found that the resulting UV-Vis profiles are largely similar, in particular for low energy transitions. In particular, the energy and orbital contributions of the optical gaps are very close.

The results support the use of the small basis set proposed here to investigate larger nanoclusters with hybrid and range-corrected functionals.

## Introduction

Monolayer-protected Nanoclusters (NCs) with a noble-metal core are particles with single-digit nanometer diameters cupped by organic ligands. They are of great interest for a great number of scientific and technological applications due to their peculiar properties in catalysis,<sup>1,2</sup> their possible interaction with biological substrates,<sup>3</sup> and their optical features.<sup>4</sup> These nanoclusters do not show metallic properties, exhibiting discrete optical gaps (usually between 1 and 2 eV) instead.<sup>5-7</sup> The production of several atomically precise and reproducible nanoclusters with definite structures, which have been determined by X-ray diffraction techniques, has recently been achieved thanks to improvements in the synthesis and crystallization techniques.<sup>8-11</sup>

In turn, the availability of accurate structures allows performing computational studies to elucidate the correlation between size, stoichiometry and optical features of these nano-objects.<sup>7,12-18</sup>

1  
2  
3  
4 Furthermore, the effect of the organic coatings can be selectively investigated too.<sup>19–21</sup>  
5  
6

7 Most synthesized monolayer-protected gold NCs are stabilized by thiolate ligands forming  
8 complex gold-thiolate interfaces as a consequence of the strong Au-S bond.<sup>22–24</sup> A number of gold  
9 clusters stabilized by phosphine ligands have been also synthesized and characterized, including  
10 the well-studied undecagold Au<sub>11</sub> clusters, the icosahedral Au<sub>13</sub> cluster coordinated by phosphine  
11 and halide ligands<sup>25,26</sup> and the [Au<sub>39</sub>(PPh<sub>3</sub>)<sub>14</sub>Cl<sub>6</sub>]Cl<sub>2</sub> cluster.<sup>27</sup>  
12  
13  
14  
15  
16

17 Gold-based NCs are often rationalized within the so called “superatom” theoretical frame-  
18 work:<sup>7</sup> with this approach, the gold core is viewed as a giant atom surrounded by almost-free  
19 electrons if the number  $n^*$  is equal to the “magic numbers” 8, 16, 20, ... The integer number  $n^*$  is  
20 given by  
21  
22  
23  
24

$$n^* = (Nv)_{\text{Au}} - W - q \quad , \quad (1)$$

25  
26  
27 with  $N$  and  $v$  the total number and the atomic valence of Au atoms, respectively,  $W$  the total  
28 number of monovalent electron-withdrawing groups bound to gold atoms, and  $q$  the overall charge  
29 of the complex in units of unsigned electrons ( $|e|$ ). This approach has been recently extended to  
30 silver-based NCs too,<sup>28</sup> but it still retains several limitations because it has a low predictive power  
31 and is limited to particles with spherical or almost-spherical noble metal cores. For example, it  
32 cannot be applied to NCs that are prolate<sup>9</sup> or are composed by more than one interacting metal  
33 core.  
34  
35  
36  
37  
38  
39  
40  
41  
42

43 Therefore, at present only an approach based on density functional theory<sup>29</sup> (DFT) and its time-  
44 dependent extensions<sup>30</sup> (TD-DFT) can provide quantitative information, in principle regardless of  
45 size and shape of the particle. While the latter point is true in theory, this approach is actually  
46 limited by the number of atoms in the systems, due to the high computational requirements needed  
47 to carry out DFT calculations. Hence, some significant approximations have to be introduced in  
48 order to study large NCs at the DFT level of theory .  
49  
50  
51  
52  
53  
54  
55

56 In the past, DFT and TD-DFT investigations have been carried out on particles ranging between  
57 11 to 102 gold atoms<sup>7,12–14,24,31–37</sup> . However, most of those studies employed plane waves as basis  
58 sets: while plane waves form an orthonormal basis, thus immune to basis set superposition errors,  
59  
60  
61  
62  
63  
64  
65

1  
2  
3  
4 in their actual implementations they only allow adopting simple functionals within the generalized  
5 gradient approximation (GGA). Only recently more advanced hybrid and range-corrected hybrid  
6 functionals have been adopted to study gold-based NCs<sup>20,21,38</sup> in combination with Gaussian basis  
7 sets.  
8  
9

10  
11  
12 In this study, we investigate the use of a small Gaussian basis set for gold atoms and ligands  
13 in five nanoclusters. Adopting it, calculations of structural properties of Au-based NCs larger  
14 than Au<sub>25</sub> and their optical response with the range-corrected hybrid CAM-B3LYP functional can  
15 become feasible.  
16  
17  
18  
19  
20  
21

## 22 Investigated Nanoclusters

23  
24  
25  
26  
27 In this paper we have studied five Au-based NCs ranging from Au<sub>11</sub> to Au<sub>38</sub>. In particular, two  
28 Au<sub>11</sub>-based NCs with the same metal core<sup>11,19</sup> but different types of ligands have been investi-  
29 gated, one Au<sub>24</sub><sup>+</sup> and a Au<sub>25</sub><sup>2+</sup> NCs<sup>9,39</sup> sharing the position of 24 gold atoms (only the central Au  
30 atoms differs), being present in the Au<sub>25</sub><sup>2+</sup> structure and lacking in Au<sub>24</sub><sup>+</sup>, and one Au<sub>38</sub> NC,<sup>40</sup> thus  
31 spanning both neutral and charged particles. The structure of the NCs simulated here are reported  
32 in Figure 1. Some of these NCs have been previously studied with DFT methods,<sup>13,20,21,36,38</sup> and  
33 here we want to compare the results obtained with a suitable small Gaussian basis set with those  
34 obtained with larger Gaussian bases. The two Au<sub>11</sub>-based NCs can be viewed as superatoms with  
35 8 (almost) free electrons, thus resembling the outer orbital shell of noble gas elements, whereas the  
36 other three cannot be rationalized as such, since they are far from being spherical.  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47

48 In order to speed up our computations, in calculations on the Au<sub>11</sub>, Au<sub>24</sub><sup>+</sup> and Au<sub>25</sub><sup>2+</sup> NCs only  
49 the metal core and the atoms directly bonded to them (saturated by H atoms in order to com-  
50 plete their valence) were retained. Thus, the aromatic thiols (SPh) and phosphines (PPh<sub>3</sub>) that  
51 protect these gold cores are largely omitted. This simplification scheme has been widely applied  
52 before<sup>7,12,24,31,32</sup> and validated at least for the cluster composed by 11-25 gold atoms.<sup>20,21,38</sup> For  
53 the Au<sub>38</sub> NC, since the gold core is capped with SC<sub>2</sub>H<sub>4</sub>Ph thiols, the ligands are substituted with  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65

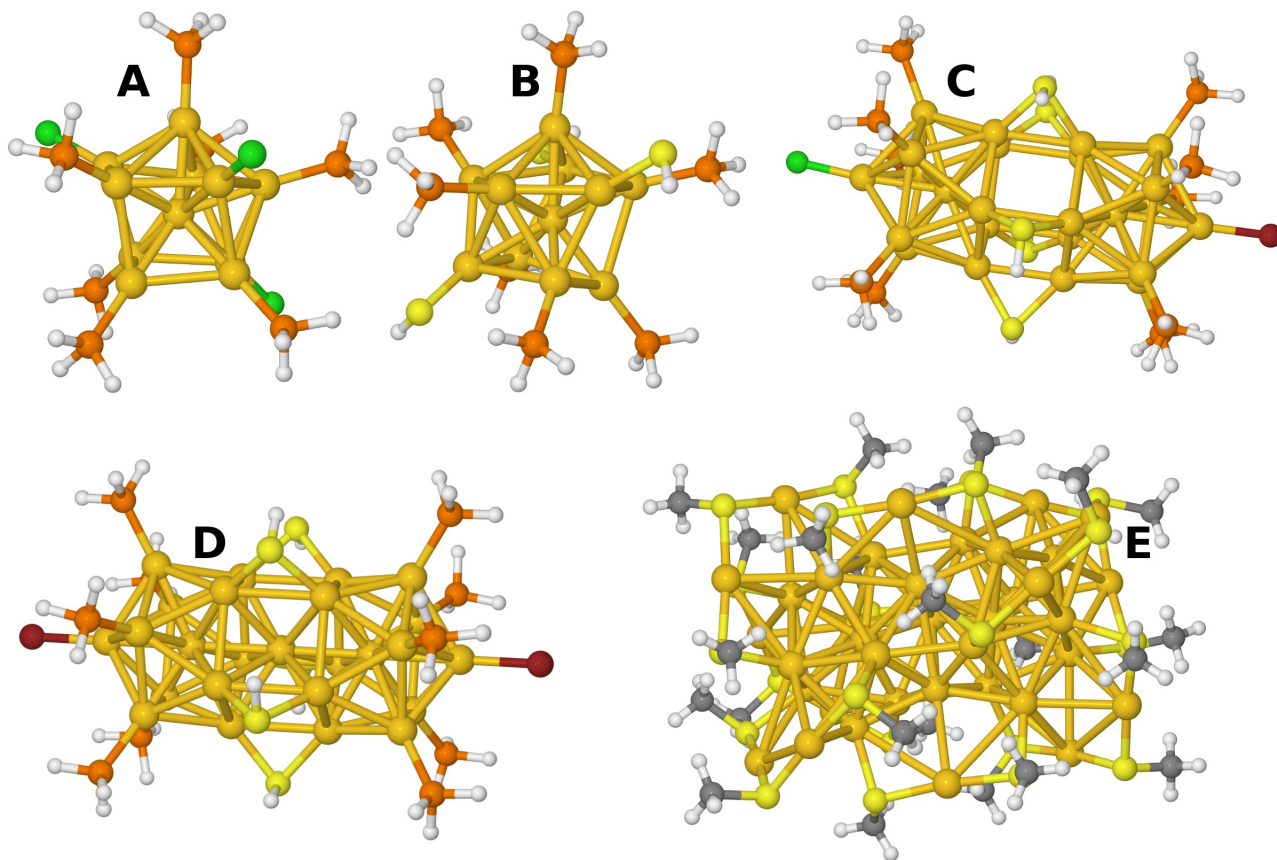


Figure 1: Nanoclusters investigated here. (A)  $\text{Au}_{11}(\text{PH}_3)_7\text{Cl}_3$ , hereafter referred to as  $\text{Au}_{11}(1)$ . (B)  $\text{Au}_{11}(\text{PH}_3)_7(\text{SH})_3$ , hereafter referred to as  $\text{Au}_{11}(2)$ . (C)  $\text{Au}_{24}(\text{PH}_3)_{10}(\text{SH})_5\text{ClBr}^+$ . (D)  $\text{Au}_{25}(\text{PH}_3)_{10}(\text{SH})_5\text{Br}_2^{2+}$ . (E)  $\text{Au}_{38}(\text{SCH}_3)_{24}$ .

S- $\text{CH}_3$  groups.

## Methods and Computational Details

All calculations were performed with the Gaussian 09 (release D) suite of programs,<sup>41</sup> in combination with the generalized gradient approximated B-PBE<sup>42,43</sup> exchange-correlation (XC) functional for structural optimization and the range separated hybrid XC functional CAM-B3LYP<sup>44</sup> for TD-DFT optical excitations, carried out within the linear response scheme. These functionals have been previous validated on gold NCs as able to correctly reproduce the structure and the UV-Vis spectra, respectively.<sup>20,21</sup> Self-consistent-field and optimization convergence criteria have been set at the “tight” level of accuracy. Optical transitions were convoluted into spectra with the Gaussview

1  
2  
3  
4 software.<sup>45</sup> Orbital isosurfaces were also plotted with the software Gaussview and structures with  
5 Jmol<sup>46</sup> adopting the standard CPK color code. The electronic density of states analysis was per-  
6 formed with the Multiwfn software.<sup>47,48</sup>  
7  
8  
9

10 We have tested a mixed combined Gaussian basis set and electronic core potential (ECP) to  
11 perform fast simulation on a range of Au-based nano-objects of different complexity, shape and  
12 charge state. Namely, we have used the small LanL2-MB<sup>49,50</sup> basis set to simulate gold atoms: this  
13 basis set adopts the LanL2 ECP for core electrons in combination with a STO-3G minimal basis  
14 set for valence electrons; atoms bonded to the metal core have been described with the 6-21G\*  
15 basis set and all other atoms with the STO-3G basis set. For reason of shortness, hereafter we refer  
16 to this mixed LanL2-MB/6-21G\*/STO-3G basis set as LanL2-MB\*.  
17  
18  
19  
20  
21  
22  
23  
24

25 This choice of basis set has been compared with calculations (geometrical optimizations with  
26 B-PBE functional) performed with different bases on the two Au<sub>11</sub>, Au<sub>24</sub><sup>+</sup>, and Au<sub>25</sub><sup>2+</sup> NCs:  
27  
28  
29

- 30 • full LanL2-MB on all atoms.
- 31
- 32
- 33 • full LanL2-DZ<sup>49,51,52</sup> on all atoms.
- 34
- 35
- 36 • modified LanL2-DZ<sup>53</sup> on gold atoms (which adds diffuse  $|p\rangle$  orbitals to LanL2-DZ), 6-  
37 21G\* on atoms bonded to the gold cores and STO-3G on all other atoms. This basis set shall  
38 be hereafter referred to as mod-LanL2-DZ\* and has been included thanks to the Basis  
39 Set Exchange database.<sup>54</sup>  
40  
41  
42  
43  
44  
45

46 The use of the modified LanL2-DZ to describe Au atoms has been suggested in Ref. 20 as one of  
47 the best ones to correctly reproduce the structural features of Au-based NCs.  
48  
49

50 We have then employed mod-LanL2-DZ\* and LanL2-MB\* bases to study the optical properties  
51 of the NCs at the TD-CAM-B3LYP level of theory. As an application, the LanL2-MB\* basis set  
52 has then been applied to the Au<sub>38</sub>-based NC to analyze its structural and optical features.  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65

# Results and Discussions

## Structural Features

The clusters have been structurally optimized with the B-PBE functional, as described in the previous section. Five different basis sets have been employed for checking their accuracy in maintaining the experimental geometry. In order to evaluate this accuracy, the atom-averaged unsigned displacement ( $\langle \delta \rangle$ ) of the relaxed structure (*opt*) with respect to the experimental structure (*exp*) has been computed as

$$\langle \delta \rangle = \langle |r_{ij}^{exp} - r_{ij}^{opt}| \rangle ,$$

where  $r_{ij}$  represents the distance between  $i$  and  $j$  metal atoms, as previously done in Ref. 20. The results of these optimizations are summarized in Table 1, while in Table 2 the time required to achieve wavefunction convergence with the various basis sets are reported for the smallest cluster.

**Table 1:** Mean unsigned error of the optimized structures of the five NCs with 4 different combination of basis sets. Values are reported in Å.

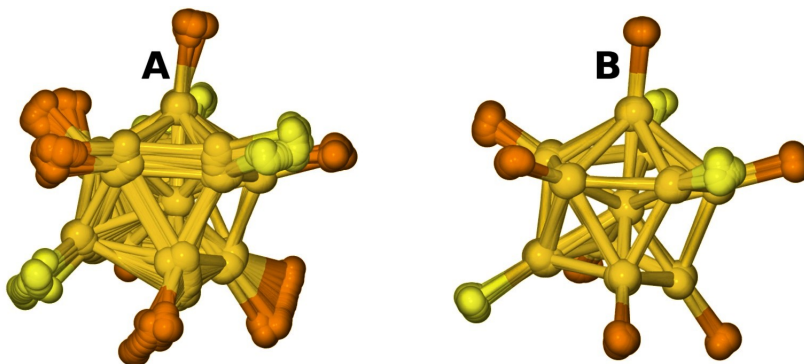
$\langle \delta \rangle / \text{Å}$	LanL2-MB	LanL2-DZ	mod-LanL2-DZ*	LanL2-MB*
Au <sub>11</sub> (1)	> 0.25	0.17	0.02	0.04
Au <sub>11</sub> (2)	> 0.25	0.16	0.02	0.03
Au <sub>24</sub> <sup>+</sup>	> 0.25	0.22	0.06	0.07
Au <sub>25</sub> <sup>2+</sup>	> 0.25	0.19	0.05	0.07
Au <sub>38</sub>	—	—	—	0.12

**Table 2:** Time to achieve self consistent field convergence (standard Gaussian 09 “tight” criteria) on a machine made up of two 8-Core Intel Xeon E5-2670 clocked at 2.60 GHz with 24 GB of random access memory and an hard disk at 7200 RPM. The time is reported in minutes.

time / min	LanL2-MB	LanL2-DZ	mod-LanL2-DZ*	LanL2-MB*
Au <sub>11</sub> (1)	8.3	17.6	27.9	9.4

As can be appreciated, the LanL2-MB basis set yields the worse structures, which is not unexpected since it is also the smallest basis set investigated here. We have to pinpoint that values

1  
2  
3  
4 of  $\langle \delta \rangle$  parameter larger than  $0.15\text{\AA}$  are associated to appreciably distorted final geometries, and  
5 values larger than  $0.25\text{\AA}$  usually mean that the network of Au-Au bonds is at least partially de-  
6  
7  
8  
9  
10  
11  $\langle \delta \rangle > 0.25$  (left) and one with  $\langle \delta \rangle \sim 0.02$  (right).



12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26 Figure 2: Superimposed experimental and relaxed structures of  $\text{Au}_{11}(2)$ . (A) Computed with the LanL2-MB basis  
27 set. (B) Computed with the mod-LanL2-DZ\* basis set.

28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65  
LanL2-DZ gives results of greater accuracy, and even more so the mod-LanL2-DZ\*, which was already proved to be one of the most accurate Gaussian basis set for gold NCs in previous works.<sup>20,21,38,55</sup> However, LanL2-MB\* yields optimized structures of fair accuracy, better than the larger LanL2-DZ. For the smallest NC studied here, namely  $\text{Au}_{11}(1)$ , we observe that LanL2-MB\* and LanL2-MB require about half the time of LanL2-DZ to achieve wavefunction convergence, which corresponds to about one third of the time required with mod-LanL2-DZ\*. Thus, overall, LanL2-MB\* seems a reasonable trade off between accuracy and the computation resources required to run the calculations.

This could be somewhat surprising since basically LanL2-MB and LanL2-MB\* differ only in the treatment of the atoms directly bonded to the gold cores (S, P, Cl, and Br atoms in the five NCs investigated here). With the pure LanL2-MB a STO-3G basis set is employed on them (from third and fourth row elements also the LanL2 ECP is also adopted), whereas with LanL2-MB\* a polarized 6-21G\* basis is used on them. To assess if the improved accuracy in the passage  $\text{STO-3G} \rightarrow 6\text{-}21\text{G}^*$  is due to the change from single to double zeta basis or to the adding of polarization functions, we also performed geometry relaxation on the two  $\text{Au}_{11}$  clusters with a

1  
2  
3  
4 basis set identical to LanL2-MB\* but with 6-21G functions on atoms bonded to the Au cores, thus  
5  
6 dropping the polarization functions. With this choice, we observed values of  $\langle \delta \rangle$  parameter  
7  
8 of 0.22 and 0.23 for Au<sub>11</sub>(1) and Au<sub>11</sub>(2) NCs, respectively, worse than what is obtained with  
9  
10 LanL2-DZ basis set and only slightly better than what obtained with LanL2-MB.

11  
12 Hence, structural optimizations can be improved not only adding diffuse functions on the gold  
13  
14 cores, as done with the modified LanL2-DZ combined ECP and basis set (as observed in Ref.s  
15  
16 20,21,38,55), but also adding polarization functions on the nearby atoms. This is most likely due  
17  
18 to the fact that Au cores are often bonded to thiols and phosphines, and both S and P atoms are  
19  
20 usually hypervalent in these cases and thus polarization functions allow them to relax towards more  
21  
22 realistic geometrical conformations.  
23  
24

## 25 26 27 **Optical and Electronic Properties**

28  
29 The electronic and optical features of the five NCs has been computed at the CAM-B3LYP and  
30  
31 TD-CAM-B3LYP level of theory with the LanL2-MB\* basis set and compared with those obtained  
32  
33 with mod-LanL2-DZ\*.  
34  
35

36  
37 First, we have investigated the charge redistribution in the NCs with the two basis sets in terms  
38  
39 of Hirshfeld partial charges,<sup>56</sup> since they are less sensitive to the basis set employed and were  
40  
41 already reported in literature.<sup>20,38</sup> It can be observed that both of them predict the same sign of  
42  
43 the charge for homologous atoms even if the two basis sets yield different partial charges. As  
44  
45 a comparison, it has to be kept in mind that similar calculations adopting the same basis set but  
46  
47 different functionals may yield for the same atom a severely positive or negative net charge (see for  
48  
49 example Ref. 20). Furthermore, both basis sets predict the same qualitative trend in the distribution  
50  
51 of charge: in particular, it has been found<sup>20,38</sup> that inner gold atoms (i.e. Au atoms bonded only  
52  
53 to other Au atoms) are significantly less negatively charged than metal atoms on the surface, a  
54  
55 behavior reproduced by both LanL2-MB\* and mod-LanL2-DZ\*. In any case, the average gold  
56  
57 charges computed with LanL2-MB\* are about 60-70% of those computed with mod-LanL2-DZ\*.  
58  
59

60 We also analyzed the energy of the first optical transition ( $S_1 \leftarrow S_0$ ), which corresponds to the  
61  
62

**Table 3: Hirshfeld partial charges in units of  $|e|$ , with  $e$  being the electron charge. Calculations were performed at the CAM-B3LYP level of theory. All data are averaged on the specified type of atoms, unless a the label “inner atom” is reported: a Au “inner atom” is a gold atom surrounded only by other Au atoms.**

	mod-LanL2-DZ*	LanL2-MB*
<b>Au<sub>11</sub> (1)</b>		
Au	-0.116	-0.085
Au (inner atom)	-0.043	-0.008
P	0.352	0.320
Cl	-0.401	-0.473
<b>Au<sub>11</sub> (2)</b>		
Au	-0.189	-0.109
Au (inner atom)	-0.059	-0.011
P	0.294	0.209
S	-0.236	-0.259
<b>Au<sub>24</sub><sup>+</sup></b>		
Au	-0.076	-0.062
Au (inner atoms)	-0.024	-0.010
Br/Cl	-0.431	-0.476
P	0.362	0.354
S	0.015	-0.034
<b>Au<sub>25</sub><sup>2+</sup></b>		
Au	-0.080	-0.045
Au (inner atoms)	-0.004	-0.008
Br	-0.314	-0.421
P	0.359	0.383
S	0.033	0.013
<b>Au<sub>38</sub></b>		
Au	—	-0.045
Au (inner atoms)	—	-0.004
S	—	0.017

so called “optical gap” of the cluster,<sup>20,57,58</sup> as reported in Table 4. As can be noticed, calculations with mod-LanL2-DZ\* and LanL2-MB\* basis sets yields very similar optical gaps, with a maximum difference of  $\sim 0.15$  eV for the Au<sub>25</sub><sup>2+</sup>-based NC; for the Au<sub>11</sub> and Au<sub>24</sub><sup>+</sup>-based NCs the difference is only  $< 0.10$  eV. Moreover, calculations performed with the LanL2-MB\* basis, despite its limitations, yield values slightly closer to the experimental data (when the latter ones are available, namely for Au<sub>11</sub>(1), Au<sub>24</sub><sup>+</sup>, and Au<sub>25</sub><sup>2+</sup> NCs). It has to be noticed that for Au<sub>24</sub><sup>+</sup> it is difficult to extrapolate the true optical gap, because of some contradicting data between optical and elec-

**Table 4: Optical gaps of the nanoclusters.**

Energy / eV	mod-LanL2-DZ*	LanL2-MB*	Exp	
Au <sub>11</sub> (1)	2.86	2.77	~2.1	Ref. 19
Au <sub>11</sub> (2)	2.62	2.64	—	
Au <sub>24</sub> <sup>+</sup>	2.05	2.04	1.3 - 1.9	Ref.s 9,20
Au <sub>25</sub> <sup>2+</sup>	2.21	2.07	1.79	Ref. 39
Au <sub>38</sub>	—	1.33	—	

trochemical measurements.<sup>9,20</sup> We have also to pinpoint that with the LanL2-MB\* basis set the energy of the optical gap appears shrinking with the increase of the particle size, something that could qualitatively be expected since larger particles should have optoelectronic properties closer to those of bulk gold rather than sub-nanometer NCs.

The  $S_1 \leftarrow S_0$  excitations can also be investigated on the basis of the most relevant orbital contributions to them, and computations with both mod-LanL2-DZ\* and LanL2-MB\* show that the LUMO $\leftarrow$ HOMO transition is the main component; Figure 3 reports the frontier orbitals for Au<sub>11</sub>(1) with the two basis sets. In the case of Au<sub>24</sub><sup>+</sup> and Au<sub>25</sub><sup>2+</sup> also the LUMO $\leftarrow$ HOMO-1 transitions give an appreciable contribution (about 40% of the LUMO $\leftarrow$ HOMO contribution), while for the Au<sub>38</sub>-based NC the LUMO+1 $\leftarrow$ HOMO-1 one gives a very important contribution (about 80% of the LUMO $\leftarrow$ HOMO).

Another important feature of these gold NCs is the energy of the first charge-transfer excitation, namely the first ligand $\leftarrow$ core band. For NCs Au<sub>11</sub>(1) and Au<sub>11</sub>(2) it was already established<sup>21</sup> that the first of such excitations occur at ~4.83 eV (LUMO+10 $\leftarrow$ HOMO) and ~4.27 eV (LUMO+6 $\leftarrow$ HOMO-2), respectively, at the TD-CAM-B3LYP level of theory with a basis set very similar to mod-LanL2-DZ\*. Even if the ligands here are mostly omitted, this comparison can still be carried out at least in a qualitatively fashion, as we report here for Au<sub>11</sub>(1). The first charge-transfer excitation with mod-LanL2-DZ\* occurs at ~5.06 eV, and is mainly of the type LUMO+10 $\leftarrow$ HOMO-1, while with LanL2-MB\* it occurs at ~4.95 eV and is mainly of the type LUMO+11 $\leftarrow$ HOMO. The orbitals involved in this transition are reported in Figure 4. While the orbitals are somewhat different, and the transition energies differs of ~0.11 eV, they are still qual-

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65

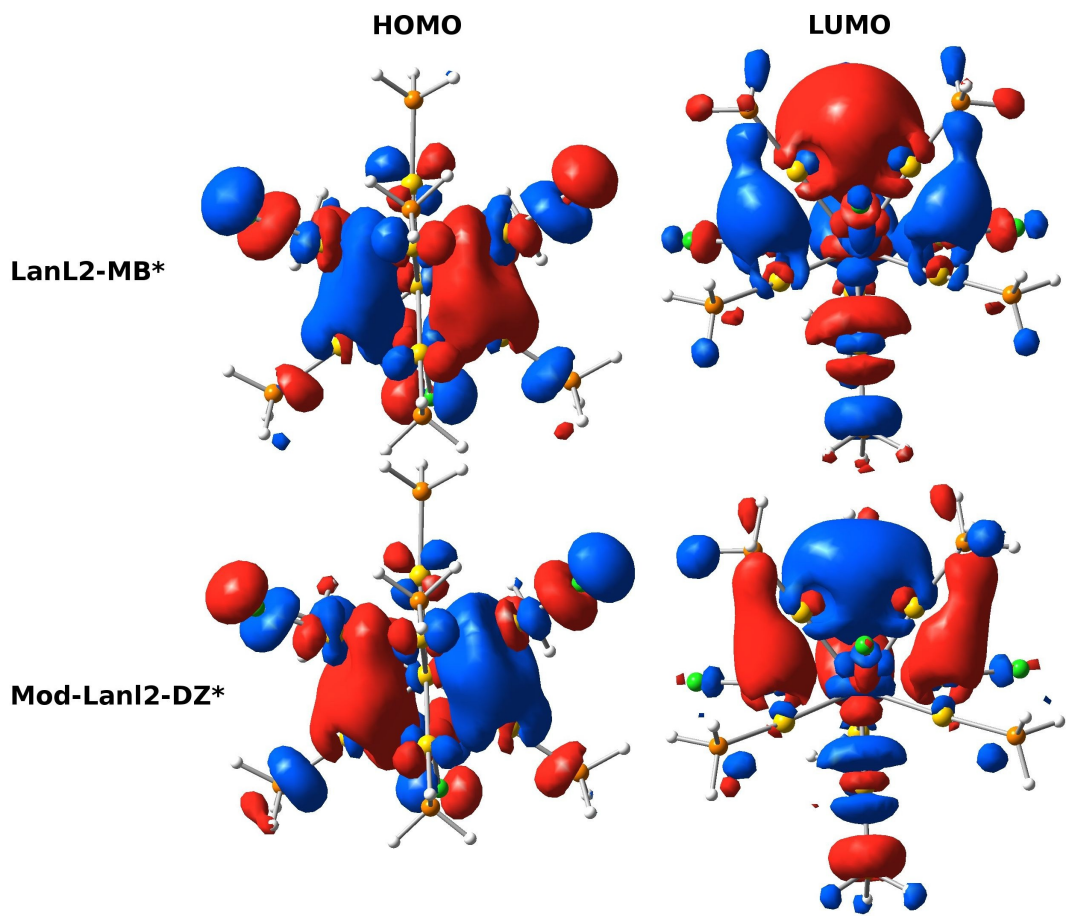


Figure 3: Isosurfaces of the frontier orbitals of Au<sub>11</sub>(1), computed with both the mod-LanL2-DZ\* and LanL2-MB\* basis sets.

itatively close.

This latter point could also be assessed by comparing the optical spectra calculated with the two basis sets, as reported in Figure 5. A redshift with the size increase is particularly evident, as discussed before about the optical gaps. For all the NCs for which a comparison is possible, the overall shapes of the spectra computed with the two basis sets seems very similar, in particular for the first low energy bands. For Au<sub>11</sub>(1) this is very noticeable. The shape yielded with LanL2-MB\* seems slightly more structured (in particular for Au<sub>11</sub>(2) and Au<sub>25</sub><sup>2+</sup>), suggesting that the LanL2-MB\* excitations are less uniformly distributed at high energy. In fact, the spike transitions are less similar, suggesting that the nest of excitations are somewhat different even if they yield very similar smoothed spectra. However, some transitions are clearly recognizable as just being red or

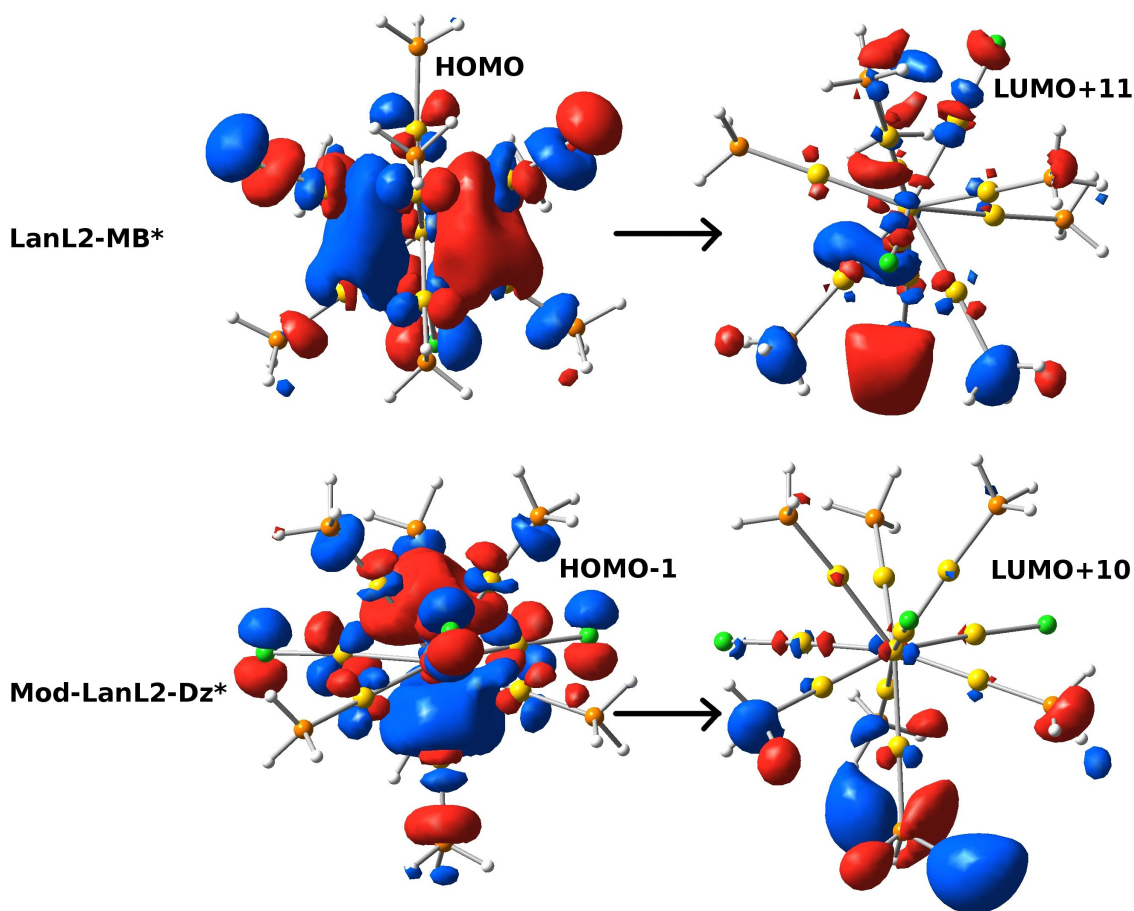


Figure 4: Isosurfaces of the orbitals involved in the first charge transfer transition (ligand←core) with the mod-LanL2-DZ\* and LanL2-MB\* basis sets of the Au<sub>11</sub>(1) NC.

blue shifted with respect to one basis set. Unfortunately, it seems not possible to predict if LanL2-MB\* excitations are red or blue-shifted with respect to those computed with the mod-LanL2-DZ\*: for example, they are blue-shifted for Au<sub>11</sub>(2) but they are red-shifted for Au<sub>25</sub><sup>2+</sup>; this impairs the possibility to systematically translate the excitations computed with LanL2-MB\* of a known factor to match the others.

The spectrum of Au<sub>38</sub> appears largely similar to that of the Au<sub>25</sub><sup>2+</sup> NC, which is reasonable since both of them are basically made up of two subunits (two Au<sub>11</sub> for Au<sub>25</sub><sup>2+</sup>, and two Au<sub>13</sub> for Au<sub>38</sub>). Figure 5. The electronic density of states (eDoS), pictured in Figure 6, displays a very small band-gap, which is qualitatively consistent with the shrinking of the optical gap with the increase in particle size. Eigenstates localized on S atoms (in yellow in Figure 6) give little contributions

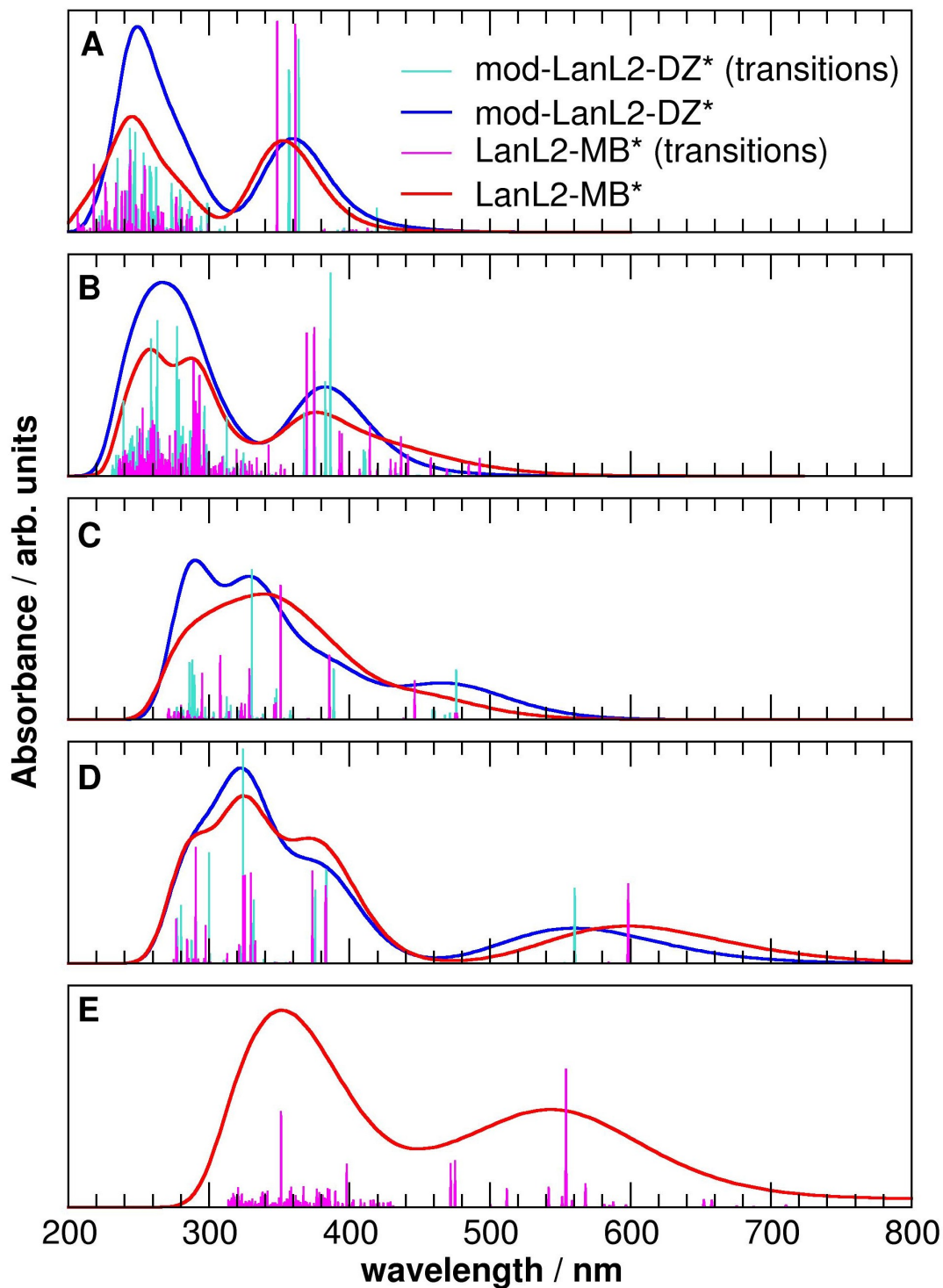


Figure 5: Computed TD-CAM-B3LYP spectra (200 excitations) of the Au<sub>11</sub>(1) (A), Au<sub>11</sub>(2) (B), Au<sub>24</sub><sup>+</sup> (C), Au<sub>25</sub><sup>2+</sup> (D), and Au<sub>38</sub> (E) NCs. In blue and red are reported the spectra smoothed with Gaussian functions with half-width at half-maximum of 0.25 eV, while in cyan and magenta the actual transitions.

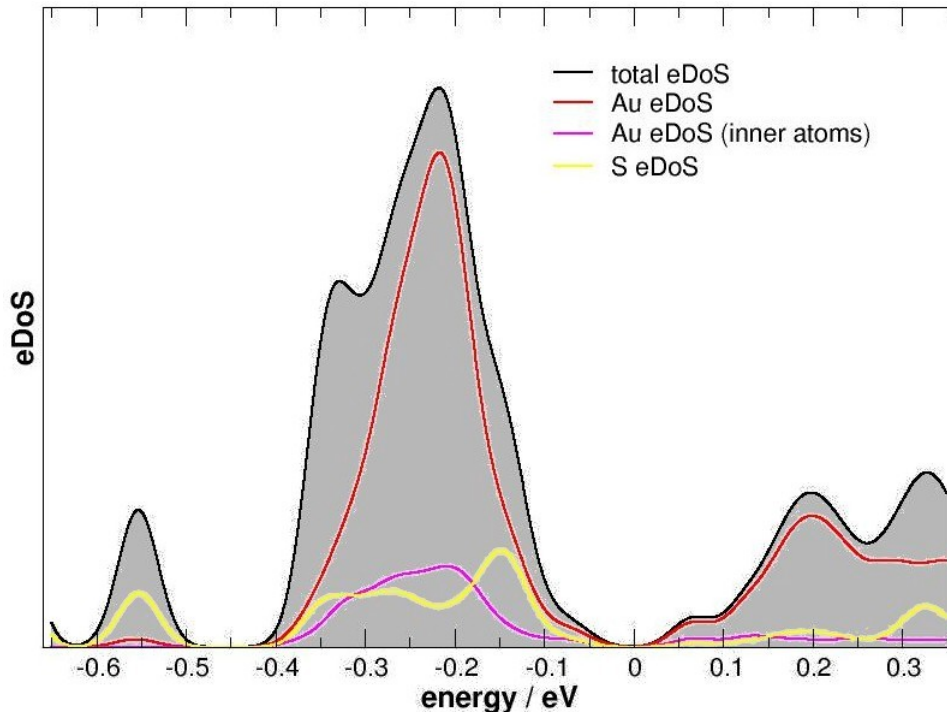


Figure 6: Electronic density of states of the Au<sub>38</sub>-based NC. The eDoS has been translated to have the band-gap at 0 eV. The eDoS has been smoothed with Gaussian functions with half-width at half-maximum of 0.05 Hartree.

to virtual orbitals close to the band-gap, while they are important to the occupied orbitals. Orbitals on the inner gold atoms (in magenta in Figure 6) give almost no contribution to virtual orbitals and are significant just for occupied orbitals distant 0.2 eV from the band-gap. Thus, the first optical excitations mainly involve surface gold atoms and not Au atoms deep into the core of the particle.

## Concluding Remarks

In this work we have carried out structural optimizations and optical spectra calculations on five monolayer protected gold-based nanoclusters.

We have employed four different basis sets during the structural optimizations with the B-PBE exchange-correlation functional, to assess which one yielded relaxed structures closer to those resolved by means of X-Ray diffraction. While it has been recovered that solutions based on the modified LanL2-DZ basis give the best accord with the experimental data, we have noted that the much smaller LanL2-MB basis set (readily available in Gaussian 09) can yield structures of similar

1  
2  
3  
4 accuracy if coupled with a polarized basis set on the atoms bonded to the gold cores (in the case  
5 of our tested nanoclusters, namely S, P, Cl). In fact, basis sets larger than LanL2-MB such as the  
6 standard LanL2-DZ fail to achieve a similar accuracy because they do not provide polarization  
7 functions on those atoms, which are often hypervalent when they are bound to Au atoms. This  
8 suggest that a good way to increase the accuracy of structural calculations on monolayer-protected  
9 gold nanoclusters is also to improve the description of the “surface” atoms. We thus propose to  
10 use the LanL2-MB basis set on Au atoms and a basis like 6-21G\* at least on S and P atoms of the  
11 cupping ligands.  
12  
13  
14  
15  
16  
17  
18  
19  
20

21 We have also carried out optical calculations at the TD-CAM-B3LYP level of theory with the  
22 basis set based on LanL2-MB and modified LanL2-DZ. Optical gaps are largely similar, both as  
23 excitation energy and orbital components. The resulting UV-Vis profiles are largely similar, in  
24 particular for the low energy transitions close to the optical gap. In this regard, orbitals close in  
25 energy to the HOMO-LUMO gap computed with one basis are basically indistinguishable from  
26 those computed with the other one. Also, charge transfer excitations of the type Ligand←Metal  
27 are recovered at very similar energies. The higher excitations, however, present some noticeable  
28 difference even if when smoothed they still yield more or less the same optical profiles.  
29  
30  
31  
32  
33  
34  
35  
36  
37

38 Therefore, we suggest the use of small basis set LanL2-MB in combination with a polarized  
39 basis set on boundary atoms in order to relax experimental structures and compute optical gaps of  
40 gold-based nanoclusters, in lieu of the much larger modified LanL2-DZ basis set. This choice can  
41 also be adopted to predict the general shape of the visible absorption bands.  
42  
43  
44  
45  
46

47 This basis set makes feasible investigating with hybrid and range-separated hybrid exchange-  
48 correlation functionals (like the CAM-B3LYP employed here) even larger nanoclusters, whose  
49 properties currently have been studied only with simpler generalized gradient approximated func-  
50 tionals and plane wave basis sets.  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65

## Acknowledgement

This work was supported by the Italian “*Ministero dell’Istruzione, dell’Università e della Ricerca*” (MIUR) through the ‘*Futuro in Ricerca*’ (FIRB) Grant RBF1248UI\_002 entitled “Novel Multiscale Theoretical / Computational Strategies for the Design of Photo and Thermo responsive Hybrid Organic-Inorganic Components for Nanoelectronic Circuits”, and the “*Programma di ricerca di rilevante interesse nazionale*” (PRIN) Grant 2010C4R8M8 entitled “Nanoscale functional Organization of (bio)Molecules and Hybrids for targeted Application in Sensing, Medicine and Biotechnology” is also acknowledged. CINECA granted computation time within the research project AUNANMR – HP10CJ027S.

## References

- (1) Campbell, C. T. *Science* **2004**, *306*, 234–235.
- (2) Cong, H.; Becker, C. F.; Elliott, S. J.; Grinstaff, M. W.; Porco, J. A. *J. Am. Chem. Soc.* **2010**, *132*, 7514–7518.
- (3) Falletta, E.; Bonini, M.; Fratini, E.; Lo Nostro, A.; Pesavento, G.; Becheri, A.; Lo Nostro, P.; Canton, P.; Baglioni, P. *J. Phys. Chem. C* **2008**, *112*, 11758–11766.
- (4) Bauld, R.; Hesari, M.; Workentin, M. S.; Fanchini, G. *Nanoscale* **2014**, *6*, 7570–7575.
- (5) Chen, S.; Ingram, R. S.; Hostetler, M. J.; Pietron, J. J.; Murray, R. W.; Schaaff, T. G.; Khoury, J. T.; Alvarez, M. M.; Whetten, R. L. *Science* **1998**, *280*, 2098–2101.
- (6) Alivisatos, A. P. *J. Phys. Chem.* **1996**, *100*, 13226–13239.
- (7) Walter, M.; Akola, J.; Lopez-Acevedo, O.; Jadzinsky, P. D.; Calero, G.; Ackerson, C. J.; Whetten, R. L.; Grönbeck, H.; Häkkinen, H. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 9157–9162.

- 1  
2  
3  
4 (8) Shichibu, Y.; Negishi, Y.; Tsukuda, T.; Teranishi, T. *J. Am. Chem. Soc.* **2005**, *127*, 13464–  
5 13465.  
6  
7  
8  
9 (9) Das, A.; Li, T.; Nobusada, K.; Zeng, Q.; Rosi, N. L.; Jin, R. *J. Am. Chem. Soc.* **2012**, *134*,  
10 20286–20289.  
11  
12  
13  
14 (10) Park, S.; Lee, D. *Langmuir* **2012**, *28*, 7049–7054.  
15  
16  
17 (11) Nunokawa, K.; Onaka, S.; Ito, M.; Horibe, M.; Yonezawa, T.; Nishihara, H.; Ozeki, T.;  
18 Chiba, H.; Watase, S.; Nakamoto, M. *J. Organomet. Chem.* **2006**, *691*, 638 – 642.  
19  
20  
21  
22 (12) Häkkinen, H.; Barnett, R. N.; Landman, U. *Phys. Rev. Lett.* **1999**, *82*, 3264–3267.  
23  
24  
25 (13) Goh, J.-Q.; Malola, S.; Häkkinen, H.; Akola, J. *J. Phys. Chem. C* **2013**, *117*, 22079–22086.  
26  
27  
28 (14) Hartmann, M. J.; Häkkinen, H.; Millstone, J. E.; Lambrecht, D. S. *J. Phys. Chem. C* DOI:  
29 10.1021/jp5125475.  
30  
31  
32  
33 (15) Muniz-Miranda, F.; Menziani, M. C.; Pedone, A. *The Journal of Physical Chemistry A* **2015**,  
34 *119*, 5088–5098, PMID: 25248052.  
35  
36  
37  
38 (16) Bousquet, B.; Cherif, M.; Huang, K.; Rabilloud, F. *J. Phys. Chem. C* **2015**, *119*, pp 4268–  
39 4277.  
40  
41  
42  
43 (17) Barcaro, G.; Sementa, L.; Fortunelli, A.; Stener, M. *J. Phys. Chem. C* **2014**, *118*, 12450–  
44 12458.  
45  
46  
47  
48 (18) Barcaro, G.; Sementa, L.; Fortunelli, A.; Stener, M. *J. Phys. Chem. C* **2014**, *118*, 28101–  
49 28108.  
50  
51  
52  
53 (19) Gutrath, B. S.; Englert, U.; Wang, Y.; Simon, U. *European Journal of Inorganic Chemistry*  
54 **2013**, *2013*, 2002–2006.  
55  
56  
57  
58 (20) Muniz-Miranda, F.; Menziani, M. C.; Pedone, A. *J. Phys. Chem. C* **2014**, *118*, 7532–7544.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 (21) Muniz-Miranda, F.; Menziani, M. C.; Pedone, A. *Phys. Chem. Chem.Phys.* **2014**, *16*, 18749–  
5 18758.  
6  
7  
8  
9 (22) Jadzinsky, P. D.; Calero, G.; Ackerson, C. J.; Bushnell, D. A.; Kornberg, R. D. *Science* **2007**,  
10 *318*, 430–433.  
11  
12  
13  
14 (23) Maksymovych, P.; Sorescu, D. C.; Yates, J. T. *Phys. Rev. Lett.* **2006**, *97*, 146103.  
15  
16  
17 (24) Häkkinen, H. *Nature Chemistry* **2012**, *4*, 443–455.  
18  
19  
20 (25) Shichibu, Y.; Kamei, Y.; Konishi, K. *Chem. Commun.* **2012**, *48*, 7559–7561.  
21  
22  
23 (26) Das, A.; Li, T.; Li, G.; Nobusada, K.; Zeng, C.; Rosi, N. L.; Jin, R. *Nanoscale* **2014**, *6*,  
24 6458–6462.  
25  
26  
27 (27) Teo, B. K.; Shi, X.; Zhang, H. *Journal of the American Chemical Society* **1992**, *114*, 2743.  
28  
29  
30 (28) Goh, J.-Q.; Akola, J. *The Journal of Physical Chemistry C* **2015**, *119*, 21165–21172.  
31  
32  
33 (29) Parr, R. G.; Yang, W. *Density-Functional Theory of Atoms and Molecules*; Oxford University  
34 Press, 1994.  
35  
36  
37  
38 (30) Runge, E.; Gross, E. K. U. *Phys. Rev. Lett.* **1984**, *52*, 997–1000.  
39  
40  
41 (31) Aikens, C. M. *J. Phys. Chem. C* **2008**, *112*, 19797–19800.  
42  
43  
44 (32) Zhu, M.; Aikens, C. M.; Hollander, F. J.; Schatz, G. C.; Jin, R. *J. Am. Chem. Soc.* **2008**, *130*,  
45 5883–5885.  
46  
47  
48  
49 (33) Provorse, M. R.; Aikens, C. M. *J. Am. Chem. Soc.* **2010**, *132*, 1302–1310.  
50  
51  
52 (34) Hulkko, E.; Lopez-Acevedo, O.; Koivisto, J.; Levi-Kalisman, Y.; Kornberg, R. D.; Petters-  
53 son, M.; Häkkinen, H. *Journal of the American Chemical Society* **2011**, *133*, 3752–3755,  
54 PMID: 21348523.  
55  
56  
57  
58 (35) Hadley, A.; Aikens, C. M. *J. Phys. Chem. C* **2010**, *114*, 18134–18138.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 (36) Lopez-Acevedo, O.; Tsunoyama, H.; Tsukuda, T.; Häkkinen, H.; Aikens, C. M. *J. Am. Chem.*  
5 *Soc.* **2010**, *132*, 8210–8218.  
6  
7  
8  
9 (37) Aikens, C. M. *J. Phys. Chem. Lett.* **2011**, *2*, 99–104.  
10  
11 (38) Muniz-Miranda, F.; Menziani, M. C.; Pedone, A. *The Journal of Physical Chemistry C* **2015**,  
12 *119*, 10766–10775.  
13  
14  
15  
16 (39) Wang, S.; Meng, X.; Das, A.; Li, T.; Song, Y.; Cao, T.; Zhu, X.; Zhu, M.; Jin, R. *Angew.*  
17 *Chem. Int. Ed.* **2014**, *53*, 2376–2380.  
18  
19  
20  
21 (40) Qian, H.; Eckenhoff, W. T.; Zhu, Y.; Pintauer, T.; Jin, R. *Journal of the American Chemical*  
22 *Society* **2010**, *132*, 8280–8281, PMID: 20515047.  
23  
24  
25  
26 (41) Frisch, M. J. et al. *Gaussian 09, Revision D.01*, Gaussian, Inc., Wallingford CT, 2013.  
27  
28  
29 (42) Becke, A. D. *Phys. Rev. A* **1988**, *38*, 3098–3100.  
30  
31  
32 (43) Perdew, J. P.; Burke, K.; Ernzerhof, M. *Phys. Rev. Lett.* **1996**, *77*, 3865–3868.  
33  
34  
35 (44) Yanai, T.; Tew, D. P.; Handy, N. C. *Chem. Phys. Lett.* **2004**, *393*, 51 – 57.  
36  
37  
38 (45) Dennington, R.; Keith, T.; Millam, J. *GaussView Version 5*, Semichem Inc. Shawnee Mission  
39 KS 2009.  
40  
41  
42  
43 (46) <http://www.jmol.org>, version: Jmol 12.2.2+dfsg-1.  
44  
45  
46 (47) Lu, T.; Chen, F. *J. Comput. Chem.* **2012**, *33*, 580–592.  
47  
48  
49 (48) Lu, T.; Chen, F. *Acta Chim. Sinica* **2011**, *69*, 2393–2406.  
50  
51  
52 (49) Dunning Jr., T. H.; Hay, P. J. *J. Chem. Phys.* **1985**, *82*, 270.  
53  
54  
55 (50) Hay, P. J.; Wadt, W. R. *J. Chem. Phys.* **1985**, *82*, 299–310.  
56  
57  
58 (51) Dunning Jr., T. H.; Hay, P. J. *J. Chem. Phys.* **1985**, *82*, 284.  
59  
60  
61  
62  
63  
64  
65

- 1  
2  
3  
4 (52) Dunning Jr., T. H.; Hay, P. J. *J. Chem. Phys.* **1985**, *82*, 299.  
5  
6  
7 (53) Couty, M.; Hall, M. B. *J. Comput. Chem.* **1996**, *17*, 1359–1370.  
8  
9  
10 (54) Schuchardt, K.; Didier, B.; Elsethagen, T.; Sun, L.; Gurumoorthi, V.; Chase, J.; Li, J.; Win-  
11 dus, T. *J. Chem. Inf. Model.* **2007**, *47*, 1045–1052.  
12  
13  
14  
15 (55) Goel, S.; Velizhanin, K. A.; Piryatinski, A.; Tretiak, S.; Ivanov, S. A. *J. Phys. Chem. Lett.*  
16 **2010**, *1*, 927–931.  
17  
18  
19  
20 (56) Hirshfeld, F. *Theor. Chim. Acta* **1977**, *44*, 129–138.  
21  
22  
23 (57) Kronik, L.; Stein, T.; Refaely-Abramson, S.; Baer, R. *J. Chem. Theory Comput.* **2012**, *8*,  
24 1515–1531.  
25  
26  
27  
28 (58) Baerends, E. J.; Gritsenko, O. V.; van Meer, R. *Phys. Chem. Chem. Phys.* **2013**, *15*, 16408–  
29 16425.  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65